

## Claims

- [c1] 1. A process for removing  $\text{SO}_2$ , NO, and  $\text{NO}_2$  from a gas stream comprising the steps of
- oxidizing at least a portion of NO in a gas stream to  $\text{NO}_2$  with an oxidizing means, followed by
  - scrubbing at least a portion of  $\text{SO}_2$ , NO, and  $\text{NO}_2$  from the gas stream with a scrubbing solution comprising ammonia, and having a pH between 6 and 8, and
  - removing at least a portion of any ammonia aerosols generated from the scrubbing step from the gas stream with an aerosol removal means.
- [c2] 2. The process of claim 1, wherein said oxidizing means is an electrical discharge reactor.
- [c3] 3. The process of claim 2, wherein said electrical discharge reactor is a dielectric barrier discharge reactor.
- [c4] 4. The process of claim 3, further comprising the step of oxidizing at least a portion of the NO to  $\text{HNO}_3$  with said dielectric barrier discharge reactor.
- [c5] 5. The process of claim 1, wherein said oxidizing step is adapted to result in a mole ratio of  $\text{SO}_2$  to NO of at least 2.5 to 1.
- [c6] 6. The process of claim 1, wherein said oxidizing step is adapted to result in a mole ratio of  $\text{SO}_2$  to  $\text{NO}_2$  of at least four to one.
- [c7] 7. The process of claim 1, said scrubbing solution comprising ammonia, ammonium sulfite, ammonium sulfate, and water, and having a pH between 6 and 8.
- [c8] 8. The process of claim 1, wherein said aerosol removal means is a wet electrostatic precipitator.
- [c9] 9. The process of claim 1, wherein said scrubbing step results in the formation

of ammonium sulfate, the process further comprising the step of withdrawing ammonium sulfate from the scrubbing solution.

[c10] 10.The process of claim 4, wherein said scrubbing step results in the formation of ammonium nitrate, the process further comprising the step of withdrawing ammonium nitrate from the scrubbing solution.

[c11] 11.A process for removing  $\text{SO}_2$ , NO,  $\text{NO}_2$ , and Hg from a gas stream comprising the steps of

- a. oxidizing at least a portion of the NO in a gas stream to  $\text{NO}_2$ , and at least a portion of the Hg in a gas stream to HgO, with an oxidizing means, followed by
- b. scrubbing at least a portion of the  $\text{SO}_2$ , NO, and  $\text{NO}_2$  from the gas stream with a scrubbing solution comprising ammonia, and having a pH between 6 and 8, and
- c. removing at least a portion of any ammonia aerosols generated from the scrubbing step, and HgO, from the gas stream with an aerosol removal means.

[c12] 12.The process of claim 11, wherein said oxidizing means is an electrical discharge reactor.

[c13] 13.The process of claim 12, wherein said electrical discharge reactor is a dielectric barrier discharge reactor.

[c14] 14.The process of claim 11, wherein said aerosol removal means is a wet electrostatic precipitator.

[c15] 15.The process of claim 11, said scrubbing solution comprising ammonia, ammonium sulfite, ammonium sulfate, and water, and having a pH between 6 and 8.

[c16] 16.The process of claim 15, wherein said scrubbing step results in the formation of ammonium sulfate, the process further comprising the step of withdrawing ammonium sulfate from the scrubbing solution.

[c17]

17. An apparatus for removing  $\text{SO}_2$ , NO, and  $\text{NO}_2$  from a gas stream comprising

- a. an oxidizing means for oxidizing at least a portion of the NO in a gas stream to  $\text{NO}_2$ , followed by
- b. a scrubber suitably adapted to scrub at least a portion of the  $\text{SO}_2$ , NO, and  $\text{NO}_2$  from the gas stream with a scrubbing solution comprising ammonia, and having a pH between 6 and 8, and
- c. an aerosol removal means for removing at least a portion of any ammonia aerosols generated by the scrubber from the gas stream.

[c18]

18. The apparatus of claim 17, wherein said oxidizing means is at least one electrical discharge reactor.

[c19]

19. The apparatus of claim 18, wherein said electrical discharge reactor is at least one dielectric barrier discharge reactor.

[c20]

20. The apparatus of claim 19, wherein said dielectric barrier discharge reactor is adapted to oxidize at least a portion of the NO to  $\text{NO}_2$  and  $\text{HNO}_3$ .

[c21]

21. The apparatus of claim 17, said scrubbing solution comprising ammonia, ammonium sulfite, ammonium sulfate, and water, and having a pH between 6 and 8.

[c22]

22. The apparatus of claim 17, wherein said aerosol removal means is at least one wet electrostatic precipitator.

[c23]

23. An apparatus for removing  $\text{SO}_2$ , NO,  $\text{NO}_2$ , and Hg from a gas stream comprising

- a. an oxidizing means for oxidizing at least a portion of the NO in a gas stream to  $\text{NO}_2$ , and at least a portion of the Hg in a gas stream to  $\text{HgO}$ , followed by
- b. a scrubber suitably adapted to scrub at least a portion of the  $\text{SO}_2$ , NO, and  $\text{NO}_2$  from the gas stream with a scrubbing solution comprising ammonia, and having a pH between 6 and 8, and

c. an aerosol removal means for removing at least a portion of any ammonia aerosols generated by the scrubber, and  $\text{HgO}$ , from the gas stream.

- [c24] 24. An apparatus for removing  $\text{SO}_2$ ,  $\text{NO}$ , and  $\text{NO}_2$  from a gas stream comprising
- a. an  $\text{NO}$  oxidizer adapted to oxidize at least a portion of the  $\text{NO}$  in a gas stream to  $\text{NO}_2$ , followed by
  - b. a scrubber adapted to scrub at least a portion of the  $\text{SO}_2$ ,  $\text{NO}$ , and  $\text{NO}_2$  from the gas stream with a scrubbing solution comprising ammonia, and having a pH between 6 and 8, and
  - c. an aerosol remover adapted to remove at least a portion of any ammonia aerosols generated by the scrubber from the gas stream.
- [c25] 25. The apparatus of claim 24, wherein said  $\text{NO}$  oxidizer is at least one electrical discharge reactor.
- [c26] 26. The apparatus of claim 25, wherein said electrical discharge reactor is at least one dielectric barrier discharge reactor.
- [c27] 27. The apparatus of claim 26, wherein said dielectric barrier discharge reactor is adapted to oxidize at least a portion of the  $\text{NO}$  to  $\text{NO}_2$  and  $\text{HNO}_3$ .
- [c28] 28. The apparatus of claim 24, said scrubbing solution comprising ammonia, ammonium sulfite, ammonium sulfate, and water, and having a pH between 6 and 8.
- [c29] 29. The apparatus of claim 24, wherein said aerosol remover is at least one wet electrostatic precipitator.